## MDCM601 2022 Exam 3

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## 1 Problems

**Problem 1.** Segment of keratin protein is shown below (carbon, green; nitrogen, blue; oxygen, red; hydrogen, white). Answer the following questions about this structure.



- 1. Which amino acid is at the N-terminus?
- 2. Which amino acid is at the C-terminus?
- 3. Considering the peptide backbone, the amino end of which amino acid is hydrogen bonded (H-bond shown as a dashed cyan line) with the carbonyl of the amino acid at the N-terminus?
- 4. The length of this hydrogen bond is shown in Angstroms. What is its length in nanometers?

- 5. Is this segment a right-handed or a left-handed helix?
- 6. This segment contains several amino acid residues that have a  $\beta$ -stereocenter. Name these amino acids?
- 7. Name an amino acid in this segment whose sidechain is negatively charged at the physiological pH.
- 8. Helices are sometimes named by indicating how many amino acids are needed to make a full circle (3.6 for  $\alpha$ -helix) followed by the number of all the backbone atoms that form the ring closed by the hydrogen bond (including the polar H). How big of a ring is formed through the drawn hydrogen bond?

**Problem 2.** Consider two strands of a  $\beta$ -sheet below and answer the following questions about this structure.



- 1. Are these strands parallel or antiparallel?
- 2. Which amino acid on the front chain is paired with the glutamate on the back chain?
- 3. The side chains in  $\beta$ -sheets are parallel or perpendicular to the plane of the sheet?
- 4. Is the lysine side chain protonated or deprotonated?
- 5. Name two different aromatic amino acids in this segment?

## Problem 3. True or false?

Histidine is in its protonated form when it coordinates cationic metals in protein structures.

**Problem 4.** Use the Ramachandran plot of myoglobin below to determine which secondary structure is F43 found in. Values for its dihedral angles are:  $\Phi = -99$  and  $\Psi = 110$ .



**Problem 5.** Structure of chloramphenicol is shown. It is a bacterial translation inhibitor. Answer the following question about its structure.



- 1. Carbon with the secondary alcohol is R or S?
- 2. Carbon with hydroxymethyl is R or S?

3. Which nucleophilic amino acid side chain can react and substitute one of the chlorides from the dichloroacetamide group?

**Problem 6.** What is the functional group through which amino acyl group is attached to the tRNA molecule?

**Problem 7.** What is the functional group that is made when amino acid is activated in reaction with ATP?

Problem 8. Consider the structure of tetracycline below.



- 1. How many phenols are there?
- 2. How many enols?
- 3. How many tertiary alcohols?

**Problem 9.** Consider the anticodon region of a tRNA molecule below.



		U	С	Α	G		
First letter	υ	UUU UUC UUA UUA UUG	UCU UCC UCA UCG	UAU UAC UAA Stop UAG Stop	UGU UGC UGA UGG Trp	U C A G	Third letter
	с	CUU CUC CUA CUG	CCU CCC CCA CCG	CAU CAC CAA CAG GIn	CGU CGC CGA CGG	U C A G	
	A	AUU AUC AUA AUG Met	ACU ACC ACA ACG	AAU AAC AAA AAG Lys	AGU }Ser AGC }Arg AGA }Arg	U C A G	
	G	GUU GUC GUA GUG	GCU GCC GCA GCG	GAU GAC GAA GAG Glu	GGU GGC GGA GGG	U C A G	

Second letter

- 1. Read off the nucleotide sequence in the 3' to 5' direction.
- 2. Translate this to a sequence on the mRNA.
- 3. What is the direction in which we read the mRNA sequence?
- 4. Use the codon chart to answer which amino acid does this tRNA carry?
- 5. Which nucleotide is it attached to?
- 6. Which atom in the tRNA is the amino acyl group attached to?
- 7. What is the non-covalent interaction between nucleotides that reinforces the parallel orientation of the bases?

**Problem 10.** Puromycin (structure shown) is very similar to the structure of amino acyl group attached to the adenine of the tRNA with one key difference.



- 1. What is the functional group in puromycin that is different from the attachment of an amino acyl to the tRNA?
- 2. Puromycin halts translation by covalently capturing the activated amino acyl tRNA bound to the ribosome. What is the nucleophilic group responsible for this capture?
- 3. Which amino acid is puromycin a methylated derivative of?

**Problem 11.** Calculate the average molecular weight of a nucleotide in tRNA if its molecular weight is 24976.7 Da and it contains 76 nucleotides.

**Problem 12.** Knowing that nucleotides in tRNA are often methylated, do you expect the average molecular weight of nucleotides in tRNA to be smaller or larger than the average molecular weight of a nucleotide calculated by knowing the molecular formula and how many nucleotides there are in a long stretch of DNA.

## 2 Solutions

- 1. Leu; Ile; Ile; 0.19; right; Ile, Thr; Glu; 13
- 2. anti-parallel; Arg; perpendicular; protonated; Tyr, Phe
- 3. False
- 4.  $\beta$ -sheet
- 5. R; R; Cys
- 6. ester
- 7. mixed anhydride
- 8. 1; 2; 2
- 9. AAG; UUC; 5' to 3'; Phe; A; O; H-bonds or  $\pi$ -stacking
- 10. amide; amine; Tyr
- 11. 328.6
- 12. larger